大同大學 106 學年度 (寒)轉學入學考試試題

考試科目:有機化學

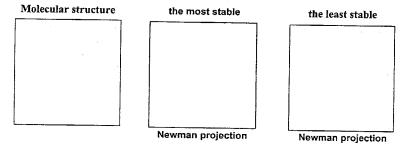
系別:化學工程學系

第以頁

註:本次考試 不可以参考自己的書籍及筆記; 不可以使用字典; 不可以使用計算器。

1. Nomenclature (15%)
(1)
(2)
(3)
(4)
(5)
Br

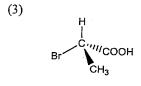
- 2. Draw the following molecular structures. (15%)
- (1) 3-ethyl-2,4,6- trimethyloctane (2) 2,2,3,4-tetramethylpentane (3) propane (4) 1-butene (5) isopropylcyclopentane
- 3. For rotation about the C-2 C-3 bond of 2,2,5,5-tetramethylhexane, please draw the molecular structure and Newman projection of its the most atable and the least stable conformers. (2+3+3%)



4. Indicate whether each of the following structures has the R or the S configuration: (16%)

For example:

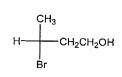
(1) (2) $\begin{array}{c} CH_2CH_2OH \\ H \longrightarrow OH \\ CH_2Br \end{array}$

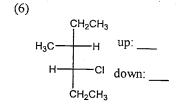


(4)

_ Right:_

Left:_





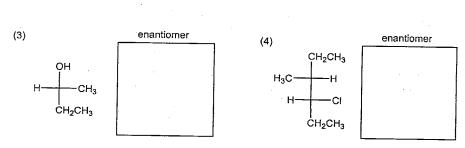
5. Draw the enantiomer of each of the following compounds: (8%) $\,$

(5)

enantiomer

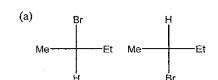
(2)

HO///



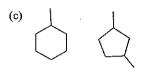
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6. For each pair of compounds below, determine whether they are identical compounds, constitutional isomers, enantiomers or diastereomers. (12%)

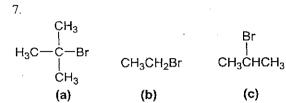




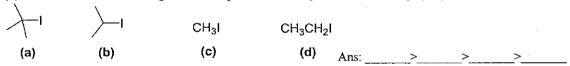




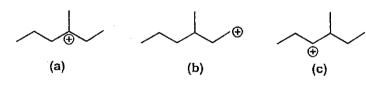
$$(\mathbf{d}) \\ \mathbf{H_3C} \overset{\mathsf{Br}}{\underset{\mathsf{CH_2CH_3}}{\mathsf{H}}} \mathbf{H} \overset{\mathsf{CH_3}}{\underset{\mathsf{Br}}{\mathsf{CH_2CH_3}}} \mathbf{H_3C} \overset{\mathsf{CH_3}}{\underset{\mathsf{Br}}{\mathsf{H}}} \mathbf{H}$$



- (1)Rank the above compounds in order of their expected reactivity toward S_N2 reaction: (6%)
- (3) Order each of the following sets of compounds with respect to E2 reactivity: (8%)



(4) Rank the following cabocations in order of decreasing stability. (6%)



Ans: _____>