大同大學 105 學年度(暑)轉學入學考試試題

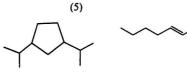
考試科目:有機化學

系别:化學工程學系性粉工程管系,第1/2頁

註:本次考試 不可以参考自己的書籍及筆記; 不可以使用字典; 不可以使用計算器。

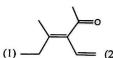
- 1. Draw the following molecular structures. (10%)
- (1) isopropyl alcohol (5) benzoic acid
- (2) 2,3,4-trimethyloctane (3) toluene (4) 2-methylhexane

- 2. Nomenclature (10%)
- (1)
- (2)
- CH₃(CH₂)₃CH₃
- (3)
- (4)



- 3. Determine whether each of the following is a <u>cis</u> or <u>trans</u> isomer: (6 %)
- (1)

- 4. Assign E or Z configuration to the following alkenes. (10%)



- 5. Indicate whether each of the following structures has the R or the S configuration: (10%) For example:

- (1)
- (2)
- (3)
- (4)
- (5)
- (6) CH₃
- (7)

- Ē Br
- OH
- HO_{II}
- ОН
- CH₂CH₃
- 6. Please draw the Newman projection of its the most stable and the least stable conformers for rotation about the C-3 —C-4 bond of 2,2,5,5tetramethylhexane. (8%)

the most stable conformer



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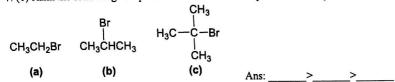
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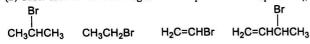
第2/2頁

註:本次考試 不可以參考自己的書籍及筆記; 不可以使用字典; 不可以使用計算器。

7. (1) Rank the following compounds in order of their expected reactivity toward E2 reaction: (3%)



(2) Order each of the following sets of compounds with respect to S_N2 reactivity: (4%)

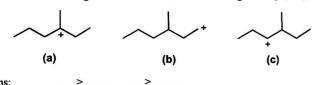


(c) (d) (a) Ans:

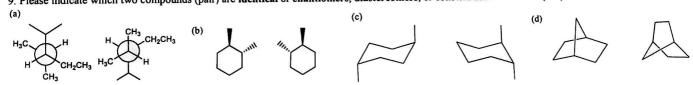
- (3) Point out which the following equations is the slowest S_N2 reaction. (4%)
- (a) $CH_3CH_2CH_3Br + HO^{\cdot} \rightarrow CH_3CH_2CH_3OH + Br^{\cdot}$ (b) $CH_3CH_2CH_2CI + HO^{\cdot} \rightarrow CH_3CH_2CH_2OH + CI^{\cdot}$

- (c) $CH_3CH_2CH_2I + HO^{-} \rightarrow CH_3CH_2CH_2OH + I^{-}$ (d) $CH_3CH_2CH_3F + HO^{-} \rightarrow CH_3CH_2CH_2OH + F^{-}$ Ans:

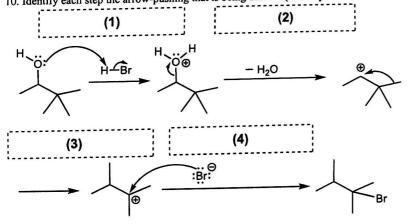
8. Rank the following cabocations in order of decreasing stability. (3%)



9. Please indicate which two compounds (pair) are identical or enantiomers, diastereomers, or constitutional isomers. (8%)



10. Identify each step the arrow-pushing that is being utilized (nucleophilic attack, loss of a leaving group, proton transfer, rearrangement). (8%)



11. Please finish the following reactions: (16%)

